



Cite this: *Chem. Commun.*, 2021, 57, 11084

Received 15th June 2021,
 Accepted 17th September 2021

DOI: 10.1039/d1cc03160k

rsc.li/chemcomm

Two-photon ESIPT-based fluorescent probe using 4-hydroxyisoindoline-1,3-dione for the detection of peroxynitrite†

Luling Wu,[†] Xue Tian,[‡] Dong Joon Lee,^b Juyoung Yoon,^c Chang Su Lim,^{*b} Hwan Myung Kim[†] and Tony D. James[†]

Excited-state intramolecular proton transfer (ESIPT)-based fluorophores with two-photon excitation fluorescence (TPEF) are rare. Our aim with this research was to develop ESIPT-based fluorophores exhibiting TPEF. Herein, we used 4-hydroxyisoindoline-1,3-dione as a scaffold to develop a two-photon fluorescent probe BHID-Bpin, for the detection of peroxynitrite (ONOO⁻). BHID-Bpin exhibits excellent selectivity, sensitivity, and fast response towards ONOO⁻ in PBS buffer solution (10 mM, pH = 7.40). Additionally, BHID-Bpin displays high photo-stability under two-photon irradiation at 750 nm. Furthermore, the probe can image endogenous ONOO⁻ in HeLa cells and exogenous ONOO⁻ in rat hippocampal slices at a depth of 110 μm.

Peroxynitrite (ONOO⁻) is formed by the combination of nitric oxide (NO) and superoxide anion radical (O₂^{•-}) under diffusion control without enzymatic catalysis.¹⁻³ It acts as a strong reactive nitrogen species (RNS) which can react with wide array of critical bioactive species, such as proteins, nucleic acids, lipids, and transition-metal enzyme centers.⁴ The concentrations of ONOO⁻ *in vivo* vary and have been estimated to be around 50–100 μM,⁵ with concentrations up to 500 μM having been reported in macrophages.⁶ In addition, hydrogen peroxide (H₂O₂) is an important reactive oxygen species (ROS), which has a key role in physiological processes, including host defense, cell proliferation, cell growth, and signaling pathways.⁷⁻⁹ Previous studies have revealed that H₂O₂ can be endogenously produced in various cells and tissues *via* the activation of NADPH oxidase.^{10,11} The exact range of H₂O₂ varies depending

on the cell line, and the intracellular homeostatic concentration ranges from 1 to approximately 100 nM.¹² Excessive production of ONOO⁻ and H₂O₂ are implicated in many pathological and physiological process associated with multiple oxidative stress-related diseases, including neurodegenerative, inflammatory, and cardiovascular diseases as well as cancer.¹³⁻¹⁶ Therefore, methods for the detection of ONOO⁻/H₂O₂ in biological systems are of considerable importance for both early diagnosis of human disease and exploration of various pathophysiologicals.

Fluorescence-based probes have been developed as powerful tools for the detection of ONOO⁻/H₂O₂, due to their good selectivity, high sensitivity, minimal invasiveness and convenience¹⁷⁻²⁴ However, traditional one-photon probes require short excitation wavelengths ranging from ultraviolet to the visible range, limiting their use in tissues or animals owing to cellular autofluorescence, and shallow tissue penetration depth.^{19,25} Over the last three decades, two-photon microscopy (TPM) has attracted particular interest since it employs two photons of low energy, near-infrared light to excite a fluorescent dye. As such TPM can overcome the above-mentioned disadvantages to a certain extent.²⁶⁻²⁹ Excited-state intramolecular proton transfer (ESIPT) fluorophores are attractive for developing reaction-based fluorescent probes. Since ESIPT fluorophores usually have large Stokes shift, high quantum yield, good photostability, and are biocompatible³⁰⁻³² Two-photon ESIPT-based probes often incorporate: 2-(2'-hydroxyphenyl)benzimidazole (HBI) or 2-(2-hydroxyphenyl)-4(3H)-quinazolinone (HPQ) as a framework. For example, two ESIPT-based fluorescent probes (1 and 2) based on HBI and HPQ respectively as the core fluorophore are illustrated in Fig. 1. James *et al.* designed probe 1 by incorporating HBI and boronate ester. The probe enabled the imaging of endogenous ONOO⁻ in RAW 264.7 macrophages and rat hippocampus tissue under two-photon excitation.³³ Zhou and colleagues developed probe 2 which used HPQ as a scaffold for the detection of H₂O₂ in HeLa cells and muscle tissues under two-photon excitation at 720 nm.³⁴ In order to extend the range of suitable ESIPT fluorophores for the construction of two-photon probes. Herein, the two-photon fluorescence properties of 2-butyl-4-hydroxyisoindoline-1,

^a Department of Chemistry, University of Bath, Bath, BA2 7AY, UK.
 E-mail: t.d.james@bath.ac.uk

^b Department of Energy Systems Research, Ajou University, Suwon 443-749, South Korea. E-mail: kimhm@ajou.ac.kr, su1222@ajou.ac.kr

^c Department of Chemistry and Nano Science, Ewha Womans University, Seoul 03760, Republic of Korea

^d School of Chemistry and Chemical Engineering, Henan Normal University, Xinxiang 453007, P. R. China

† Electronic supplementary information (ESI) available. See DOI: 10.1039/d1cc03160k

‡ These authors contributed equally to this work.



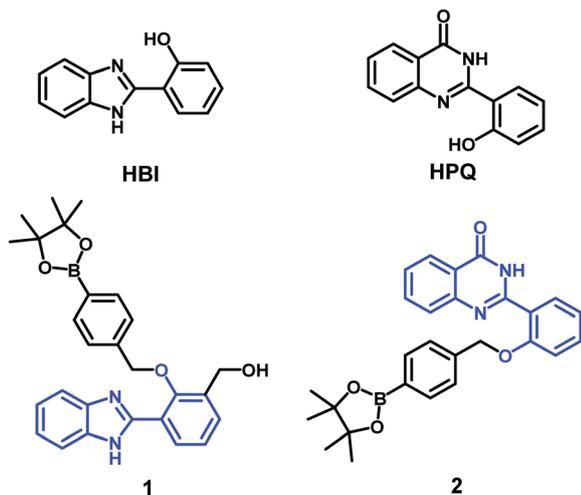
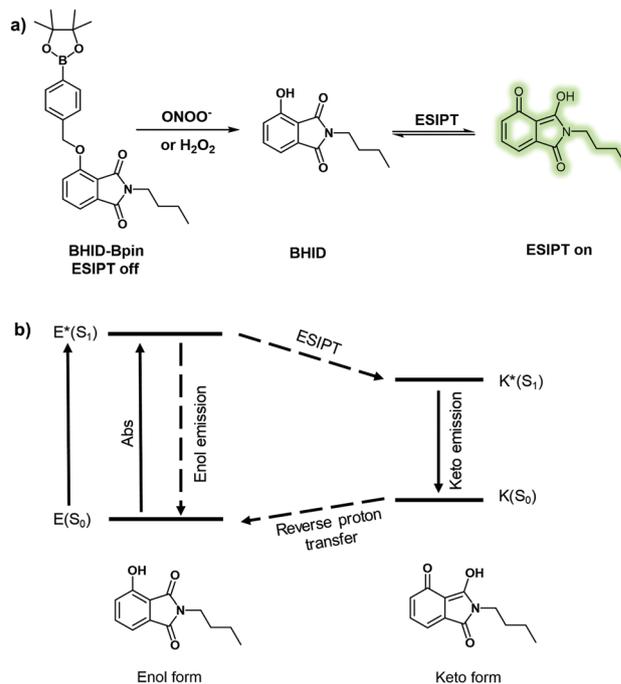


Fig. 1 Structures of previously reported two-photon ESPT-based fluorescent probes. Where HBI and HPQ are used as the fluorophores.

3-dione (**BHID**, a 4-hydroxyisoindoline-1,3-dione (HID) derivative) are explored in the development of probe **BHID-Bpin**.

BHID-Bpin can be readily prepared and is non-fluorescent. However, upon addition of $\text{ONOO}^-/\text{H}_2\text{O}_2$, the phenylboronic acid pinacol ester is oxidized by $\text{ONOO}^-/\text{H}_2\text{O}_2$ which leads to the release of phenol moiety (Scheme 1). The **BHID** unit is fluorescent, since the ESPT process is restored. The boronate pinacol ester can respond to both ONOO^- and H_2O_2 . However, it is much more sensitive to reaction with ONOO^- than H_2O_2 , because the enhanced nucleophilicity of ONOO^- leads to a reaction rate $\sim 10^6$ times faster than that for H_2O_2 .³⁵ Furthermore, different concentrations of $\text{ONOO}^-/\text{H}_2\text{O}_2$ also affects the selectivity.²² Given the possible dual reactivity, we evaluated the concentration-dependent and time-dependent reactivity of **BHID-Bpin** towards both H_2O_2 and ONOO^- in aqueous solutions. In addition, the probe was used to image ONOO^- in cells and tissues with satisfactory results, demonstrating its practical applicability.

BHID-Bpin was synthesized using a simple two-step sequence (Scheme S1, ESI[†]). Initially, **BHID** was synthesized utilizing commercially available 3-hydroxyphthalic anhydride and butylamine. Then a mixture of **BHID**, 4-bromomethylphenylboronic acid pinacol ester, and K_2CO_3 in dry DMF afforded **BHID-Bpin** in 61% yield. It is worth noting that 3-hydroxyphthalic anhydride has two sites that can be functionalized. The phenol which serves as a unit that can be used to block the ESPT process upon attachment of a receptor. Additionally its anhydride unit can be used to construct dual-responsive probes or organelle-targeting probes by varying the amine used. With **BHID-Bpin** in hand we then evaluated the optical properties of the probe in PBS buffer solution (10 mM, pH = 7.40). The probe **BHID-Bpin** (10 μM) exhibits no absorption peak, however, upon addition of ONOO^- (20 μM)/ H_2O_2 (95 μM), the probe displays a weak absorption at about 400 nm (Fig. S1 and S6, ESI[†]). Therefore, excitation at 400 nm was used to investigate the fluorescence response of the probe towards $\text{ONOO}^-/\text{H}_2\text{O}_2$ (Fig. 2 and 3).



Scheme 1 (a) Reaction between probe **BHID-Bpin** and $\text{ONOO}^-/\text{H}_2\text{O}_2$. (b) Diagram of ESPT process of **BHID**.

As expected, **BHID-Bpin** displayed weak fluorescence due to the ESPT process being blocked by the boronate ester moiety. However, with incremental addition of ONOO^- (0–20 μM)/ H_2O_2 (0–95 μM), the fluorescence spectra of **BHID-Bpin** displayed a continuous increase in emission intensity at 515 nm (Fig. 2a and 3a), that reached a plateau when 2 equiv. of $\text{ONOO}^-/9.5$ equiv. of H_2O_2 were added, causing a 31/29-fold enhancement in the fluorescence emission intensity (Fig. S2 and S7, ESI[†]). Plots of the emission intensity at 515 nm *versus* concentration of $\text{ONOO}^-/\text{H}_2\text{O}_2$ were used to determine the limit of detection to be 73 nM for ONOO^- and 221 nM for H_2O_2 (linear equation: $y_1 = 448.67 + 1000.74 \times [\text{ONOO}^-] (\mu\text{M})$, $R^2 = 0.999$, y_1 is the intensity at 515 nm, Fig. 2b; $y_2 = 632.99 + 331.02 \times [\text{H}_2\text{O}_2] (\mu\text{M})$, $R^2 = 0.986$, y_2 is the intensity at 515 nm, Fig. 3b). To determine the selectivity towards ONOO^- and H_2O_2 , various interfering species were evaluated. As shown in Fig. S3 (ESI[†]), **BHID-Bpin** was particularly responsive towards ONOO^- over H_2O_2 , ROO^\cdot ,

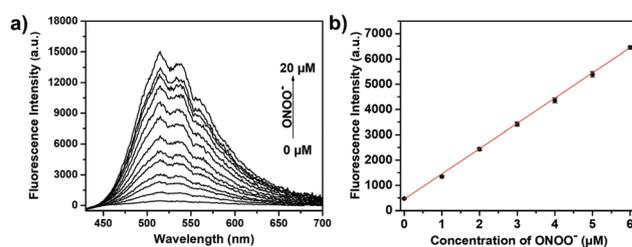


Fig. 2 (a) One-photon fluorescence spectra of **BHID-Bpin** (10 μM) after the addition of various concentrations of ONOO^- (0–20 μM) in PBS buffer solution (10 mM, pH = 7.40) after 1 min. $\lambda_{\text{ex}} = 400$ nm (b) Linear relationship between fluorescence intensity of **BHID-Bpin** (10 μM) and concentration of ONOO^- (0–6 μM) after 1 min. $\lambda_{\text{ex/em}} = 400/515$ nm.



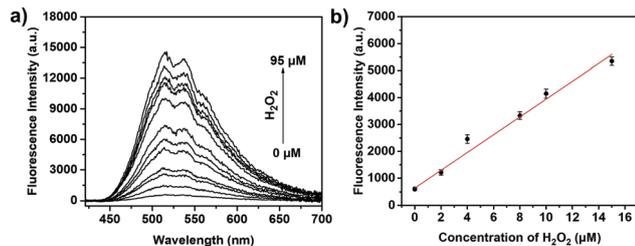


Fig. 3 (a) One-photon fluorescence spectra of **BHID-Bpin** (10 μM) with addition of various concentrations of H_2O_2 (0–95 μM) in PBS buffer solution (10 mM, pH = 7.40) after 80 min. $\lambda_{\text{ex}} = 400$ nm. (b) Linear relationship between fluorescence intensity of **BHID-Bpin** (10 μM) and concentration of H_2O_2 (0–15 μM) after 80 min. $\lambda_{\text{ex/em}} = 400/515$ nm.

$\cdot\text{OH}$, $^1\text{O}_2$, and ClO^- when the solutions were incubated for 1 min. However, the existence of ONOO^- or H_2O_2 could induce significant fluorescence enhancements of the probe **BHID-Bpin** when the solutions were incubated for 80 min and compared with other interfering species. The dynamic behaviour of probe **BHID-Bpin** was recorded in the presence of $\text{ONOO}^-/\text{H}_2\text{O}_2$. As depicted in Fig. S5 and S9 (ESI †). Upon addition of 2 equiv. of $\text{ONOO}^-/9.5$ equiv. of H_2O_2 , a fluorescence enhancement was observed with maximum fluorescence observed at 1/80 min. These results indicated that probe **BHID-Bpin** has a much faster response and was more sensitive for ONOO^- than H_2O_2 .

To evaluate the practicability of probe **BHID-Bpin** under physiological conditions, the effect of pH on the fluorescence response of the probe in the absence and presence of $\text{ONOO}^-/\text{H}_2\text{O}_2$ at various pH values was determined. The fluorescence intensities of probe **BHID-Bpin** changed over a pH range from 3.00 to 12.00. In the presence of $\text{ONOO}^-/\text{H}_2\text{O}_2$, there were notable changes in fluorescence of probe **BHID-Bpin** over a pH range from 7.40 to 9.00, (Fig. S4 and S8, ESI †) which indicated that the probe could be used to monitor $\text{ONOO}^-/\text{H}_2\text{O}_2$ under physiological conditions. The fluorescence intensity at 515 nm decreased significantly when the pH was higher than 10, probably due to inhibition of proton transfer by deprotonation of phenol groups. Decomposition of ONOO^- in an acidic environment is likely responsible for the decreased emission intensities at low pH.³⁶ In addition, LC-MS experiments were performed to confirm the proposed fluorescence mechanism (Fig. S10–S13, ESI †).

Furthermore, the two-photon action cross section ($\Phi\delta$) of **BHID-Bpin** and **BHID** was measured using rhodamine-6G as a reference (Fig. S14, ESI †).³⁷ The maxima $\Phi\delta$ value of **BHID-Bpin** and **BHID** were calculated to be 0.7 GM and 1.9 GM, respectively, under two-photon excitation at 750 nm. Moreover, a cell counting kit-8 (CCK-8) method using HeLa cells was performed with different concentrations of **BHID** and **BHID-Bpin** to determine the cytotoxicity. As shown in Fig. S15 (ESI †), both **BHID** and **BHID-Bpin** had negligible impact on cell survival. These data indicate that **BHID-Bpin** is suitable for measuring ONOO^- concentrations in a cellular environment. In addition, no photobleaching occurred after **BHID-Bpin** was irradiated for 10 min under two-photon excitation at 750 nm, which implied high photo-stability of the probe for real-time imaging (Fig. S16, ESI †). Next, **BHID-Bpin** was used to image exogenous ONOO^- and H_2O_2

respectively using the same concentration (300 μM) and incubation time (20 min) in HeLa cells. The two-photon imaging indicates that **BHID-Bpin** has higher sensitivity towards ONOO^- over H_2O_2 (Fig. S17, ESI †). We then evaluated the use of probe **BHID-Bpin** for the detection ONOO^- using other stimulus in live cells using TPM. TPM images of probe **BHID-Bpin** (5 μM) labelled HeLa cells exhibit weak fluorescence intensity (Fig. S18a, ESI †). The two-photon excitation fluorescence (TPEF) intensity was markedly increased when the cells are treated with 300 μM ONOO^- (Fig. S18b, ESI †). In addition, an increase in the intensity was observed when 3-morpholinopyridone (SIN-1), a well-known ONOO^- donor, was added (Fig. S18c, ESI †).^{38,39} However, the fluorescence decreased upon pre-treatment with aminoguanidine (an inhibitor of nitric oxide synthase)⁴⁰ or ebselen (an organoselenium compound, which is a known ONOO^- scavenger that rapidly catalyzes the reduction of ONOO^-) (Fig. S18e and f respectively, ESI †).⁴¹ **BHID-Bpin** was then used to measure endogenous ONOO^- . Using 50 ng mL^{-1} interferon- γ (IFN- γ) and 500 ng mL^{-1} lipopolysaccharide (LPS) resulted in the production of endogenous ONOO^- ,^{42,43} leading to a significant enhancement in fluorescence, compared to the control group (Fig. S18, ESI †). Next, we determined whether probe **BHID-Bpin** can be used to detect ONOO^- in mice tissue (Fig. 4). TPM images of a portion of **BHID-Bpin**-labelled fresh rat hippocampal slices were obtained. The accumulated image at depths of 90–160 μm exhibited weak fluorescence intensity (Fig. 4a); however, the TPEF intensity of the image was markedly increased when the tissues were pre-treated with 50 μM SIN-1 for 20 min (Fig. 4b). The images collected at a higher magnification clearly indicate the ONOO^- levels in the individual cells (Fig. 4c–e). In addition, the TPEF intensity decreased upon treatment with 150 μM ebselen for 40 min

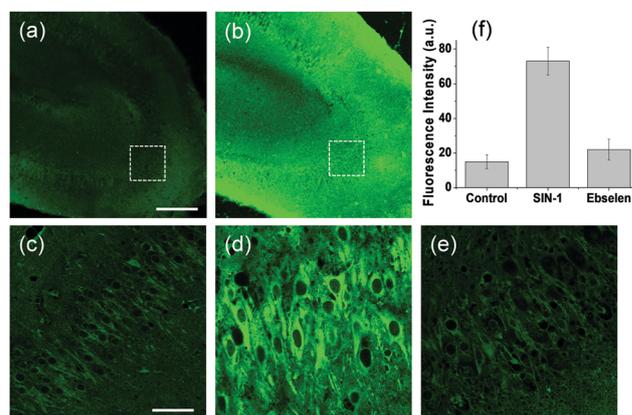


Fig. 4 TPM rat hippocampal slice images acquired after incubation of 20 μM **BHID-Bpin** for 1 h and obtained at $\times 40$ magnification. (a and b) The TPM images are implemented by combining images in the z-direction, which correspond to depths of 90–160 μm and detected in the (a) absence and (b) presence of 50 μM SIN-1. (c–e) Enlarged images show a white box part of panels (a) and (b) at a depth of 110 μm and were acquired (c) before and (d and e) after the addition of (d) SIN-1 (50 μM) for 20 min, (e) 150 μM ebselen with 50 μM SIN-1 for 40 min. (f) Average TPEF intensity in panels c–e. Images were acquired at the range of 400–600 nm upon excitation at 750 nm. Scale bar: (a) 300 and (c) 50 μm , respectively.



(Fig. 4e). These results confirm that **BHID-Bpin** can be used as a tool for the imaging of exogenous ONOO⁻ in living systems.

In summary, we have developed a fluorescent probe, **BHID-Bpin**, where HID was used for the first time as part of an ESIPT framework to detect ONOO⁻ using TPM. Our approach used a simple two-step synthetic route to prepare the two-photon activatable probe. More importantly, the probe was used to measure ONOO⁻ in HeLa cells with low cytotoxicity. Furthermore, the probe exhibited good two-photon properties and could be used to image exogenous ONOO⁻ in rat hippocampal slices at a depth up to 110 μm. Thus, our probe can be regarded as a powerful imaging tool to investigate ONOO⁻ in living biological systems. The success of **BHID-Bpin** in imaging ONOO⁻ using two-photon excitation indicates that HID is suitable for the construction of novel ESIPT-based two-photon fluorescent probes. Finally, we anticipate that the short emission wavelengths of **BHID** can be overcome by employing **BHID** as a Förster resonance energy transfer donor.⁴⁴

LW wishes to thank China Scholarship Council and the University of Bath for supporting his PhD work in the UK. XT wishes to thank the University of Bath for supporting her PhD work in the UK. We would like to thank the EPSRC and the University of Bath for funding. DJL, CSL and HMK thank the National Leading Research Lab Program of the National Research Foundation of Korea (NRF), funded by the Korean government (MSIP) (NRF-2019R1A2B5B03100278, NRF-2021R1C1C1013187), Center for Convergence Research of Neurological Disorders (NRF-2019R1A5A2026045). TDJ wishes to thank the Royal Society for a Wolfson Research Merit Award and the Open Research Fund of the School of Chemistry and Chemical Engineering, Henan Normal University for support (2020ZD01). NMR and Mass Spectroscopic (MS) Characterisation facilities were provided by the Material and Chemical Characterisation Facility (MC²) at the University of Bath (<http://go.bath.ac.uk/mc2>). All data supporting this study are provided as ESI† accompanying this paper.

Conflicts of interest

There are no conflicts to declare.

Notes and references

- J. S. Beckman, T. W. Beckman, J. Chen, P. A. Marshall and B. A. Freeman, *Proc. Natl. Acad. Sci. U. S. A.*, 1990, **87**, 1620–1624.
- R. Radi, *J. Biol. Chem.*, 2013, **288**, 26464–26472.
- G. L. Squadrito and W. A. Pryor, *Chem.-Biol. Interact.*, 1995, **96**, 203–206.
- G. Ferrer-Sueta, N. Campolo, M. Trujillo, S. Bartesaghi, S. Carballal, N. Romero, B. Alvarez and R. Radi, *Chem. Rev.*, 2018, **118**, 1338–1408.
- C. Szabó, H. Ischiropoulos and R. Radi, *Nat. Rev. Drug Discovery*, 2007, **6**, 662–680.
- O. Stachowiak, M. Dolder, T. Wallimann and C. Richter, *J. Biol. Chem.*, 1998, **273**, 16694–16699.
- M. Reth, *Nat. Immunol.*, 2002, **3**, 1129–1134.
- R. S. Balaban, S. Nemoto and T. Finkel, *Cell*, 2005, **120**, 483–495.
- M. Ushio-Fukai, *Cardiovasc. Res.*, 2006, **71**, 226–235.
- J. D. Lambeth, *Nat. Rev. Immunol.*, 2004, **4**, 181–189.
- M. T. Quinn, M. C. B. Ammons and F. R. DeLeo, *Clin. Sci.*, 2006, **111**, 1–20.
- H. Sies, *Redox Biol.*, 2017, **11**, 613–619.
- P. Pacher, J. S. Beckman and L. Liaudet, *Physiol. Rev.*, 2007, **87**, 315–424.
- G. Ferrer-Sueta and R. Radi, *ACS Chem. Biol.*, 2009, **4**, 161–177.
- C. Szabó, H. Ischiropoulos and R. Radi, *Nat. Rev. Drug Discovery*, 2007, **6**, 662–680.
- X. Bai, K. K.-H. Ng, J. J. Hu, S. Ye and D. Yang, *Annu. Rev. Biochem.*, 2019, **88**, 605–633.
- L. Wu, J. Liu, P. Li, B. Tang and T. D. James, *Chem. Soc. Rev.*, 2021, **50**, 702–734.
- L. Wu, A. C. Sedgwick, X. Sun, S. D. Bull, X.-P. He and T. D. James, *Acc. Chem. Res.*, 2019, **52**, 2582–2597.
- X. Tian, L. C. Murfin, L. Wu, S. E. Lewis and T. D. James, *Chem. Sci.*, 2021, **12**, 3406–3426.
- D. Cheng, W. Xu, X. Gong, L. Yuan and X.-B. Zhang, *Acc. Chem. Res.*, 2021, **54**, 403–415.
- Y. Li, X. Xie, X. Yang, M. Li, X. Jiao, Y. Sun, X. Wang and B. Tang, *Chem. Sci.*, 2017, **8**, 4006–4011.
- L. Wu, J. Huang, K. Pu and T. D. James, *Nat. Rev. Chem.*, 2021, **5**, 406–421.
- J.-S. Hu, C. Shao, X. Wang, X. Di, X. Xue, Z. Su, J. Zhao, H.-L. Zhu, H.-K. Liu and Y. Qian, *Adv. Sci.*, 2019, **6**, 1900341.
- X. Xie, F. Tang, G. Liu, Y. Li, X. Su, X. Jiao, X. Wang and B. Tang, *Anal. Chem.*, 2018, **90**, 11629–11635.
- G. Maulucci, G. Bačić, L. Bridal, H. H. Schmidt, B. Tavittian, T. Viel, H. Utsumi, A. S. Yalçın and M. De Spirito, *Antioxid. Redox Signaling*, 2016, **24**, 939–958.
- F. Helmchen and W. Denk, *Nat. Methods*, 2005, **2**, 932–940.
- W. R. Zipfel, R. M. Williams and W. W. Webb, *Nat. Biotechnol.*, 2003, **21**, 1369–1377.
- H. M. Kim and B. R. Cho, *Chem. Rev.*, 2015, **115**, 5014–5055.
- Y. L. Pak, S. J. Park, D. Wu, B. Cheon, H. M. Kim, J. Bouffard and J. Yoon, *Angew. Chem., Int. Ed.*, 2018, **57**, 1567–1571.
- M. Nakazono, K. Saita, C. Kurihara, S. Nanbu and K. Zaitsu, *J. Photochem. Photobiol., A*, 2009, **208**, 21–26.
- W. Feng, S. Feng and G. Feng, *Anal. Chem.*, 2019, **91**, 8602–8606.
- A. C. Sedgwick, L. Wu, H.-H. Han, S. D. Bull, X.-P. He, T. D. James, J. L. Sessler, B. Z. Tang, H. Tian and J. Yoon, *Chem. Soc. Rev.*, 2018, **47**, 8842–8880.
- M. L. Odyniec, S.-J. Park, J. E. Gardiner, E. C. Webb, A. C. Sedgwick, J. Yoon, S. D. Bull, H. M. Kim and T. D. James, *Chem. Sci.*, 2020, **11**, 7329–7334.
- L. Zhou, Y. Peng, Q. Wang and Q. Lin, *J. Photochem. Photobiol., B*, 2017, **167**, 264–268.
- A. Sikora, J. Zielonka, M. Lopez, J. Joseph and B. Kalyanaraman, *Free Radical Biol. Med.*, 2009, **47**, 1401–1407.
- M. Kirsch, H.-G. Korth, A. Wensing, R. Sustmann and H. de Groot, *Arch. Biochem. Biophys.*, 2003, **418**, 133–150.
- M. A. Albota, C. Xu and W. W. Webb, *Appl. Opt.*, 1998, **37**, 7352–7356.
- X. Li, R.-R. Tao, L.-J. Hong, J. Cheng, Q. Jiang, Y.-M. Lu, M.-H. Liao, W.-F. Ye, N.-N. Lu, F. Han, Y.-Z. Hu and Y.-H. Hu, *J. Am. Chem. Soc.*, 2015, **137**, 12296–12303.
- S. F. Peteu, R. Boukherroub and S. Szunerits, *Biosens. Bioelectron.*, 2014, **58**, 359–373.
- R. B. R. Muijsers, E. van den Worm, G. Folkerts, C. J. Beukelman, A. S. Koster, D. S. Postma and F. P. Nijkamp, *Br. J. Pharmacol.*, 2000, **130**, 932–936.
- A. Daiber, M.-H. Zou, M. Bachschmid and V. Ullrich, *Biochem. Pharmacol.*, 2000, **59**, 153–160.
- N. M. Iovine, S. Pursnani, A. Voldman, G. Wasserman, M. J. Blaser and Y. Weinrauch, *Infect. Immun.*, 2008, **76**, 986–993.
- A. H. Ding, C. F. Nathan and D. J. Stuehr, *J. Immunol.*, 1988, **141**, 2407–2412.
- L. Wu, C. Huang, B. P. Emery, A. C. Sedgwick, S. D. Bull, X.-P. He, H. Tian, J. Yoon, J. L. Sessler and T. D. James, *Chem. Soc. Rev.*, 2020, **49**, 5110–5139.

